



Solid-phase synthesis and utilization of side-chain reactive unnatural amino acids

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Abstract—Alkylation of the benzophenone imine of glycine Wang resin with α,ω -dihaloalkanes yielded valuable reactive intermediates. These racemic ω -chloro or ω -bromo intermediates were converted to α -amino acids containing diverse side-chain functionalities (e.g. ω -chlorides, nitriles, and thioethers), proline and its ring homologs, and 1-aminocycloalkancarboxylic acid derivatives.

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Over the years, non-proteinogenic, unnatural α -amino acids have been widely used as components of peptides to enhance biological activity, proteolytic stability, and bioavailability.¹ Unique conformational constraints can also be induced into peptides or peptidomimetics by introducing an appropriate amino acid precursor. Unnatural amino acids are generally prepared by synthetic routes involving solution-phase techniques, and subsequently incorporated into a peptide sequence by solid-phase methods. We have recently published methods for the preparation of resin-bound α,α -disubstituted unnatural amino acids with diverse side-chain substitutions,² and α -substituted prolines and homologs.³ This paper describes the solid-phase preparation of racemic α -monosubstituted amino acids with remote electrophilic centers. This is accomplished by alkylation of the Schiff base of a resin-bound glycinate with α,ω -dihaloalkanes.⁴ The resulting intermediates can then be converted on-resin to unnatural α -amino acids of three different types: (i) amino acids with a variety of side-chain functionalities (**4** and **6**), (ii) proline and homologs⁵ (**8**), and (iii) 1-aminocycloalkancarboxylic acid derivatives⁵ (**10**) (Scheme 1).

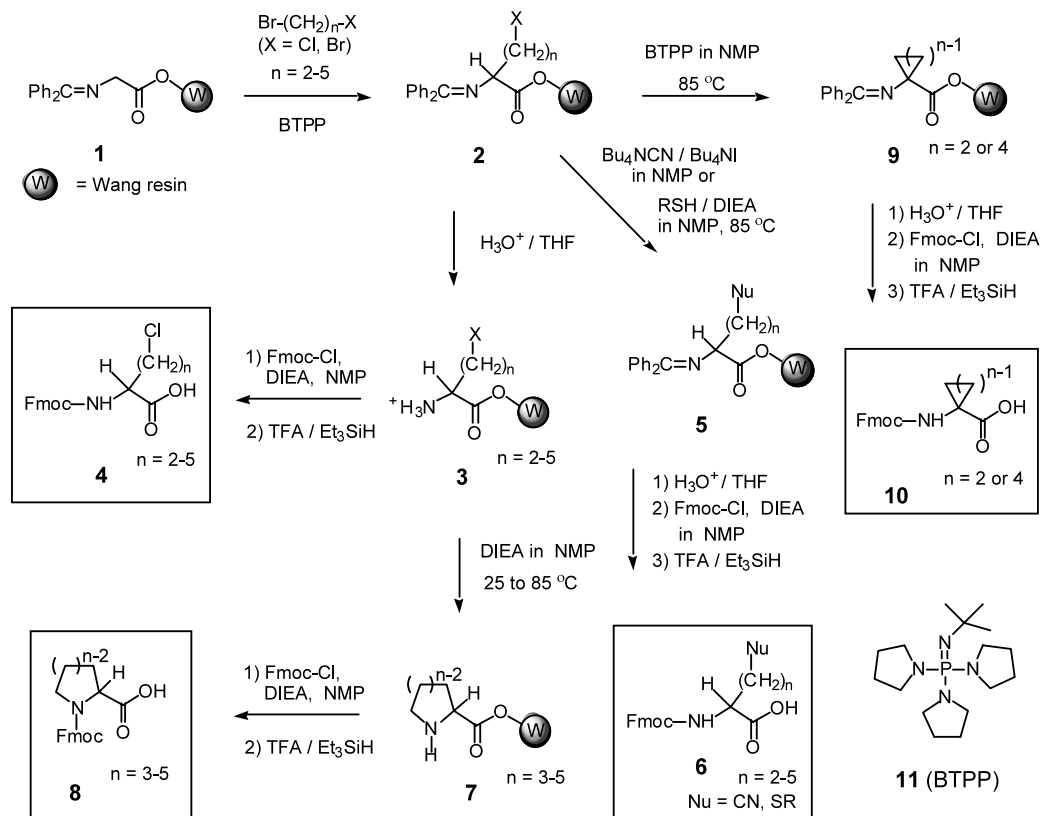
Alkylation of the activated benzophenone imine of glycine Wang resin **1** with α,ω -dihaloalkanes of different chain lengths ($n=2-5$) provided key resin-bound

racemic intermediates **2**. The alkylation of the imine was carried out in NMP under mild reaction conditions using the nonionic Schwesinger base, BTTP⁶ (**11**, 10 equiv.). In most cases, α -bromo- ω -chloro electrophiles (10 equiv.) were used for the alkylation in order to reduce side reactions such as alkene formation by elimination, premature cyclization on the nitrogen, or cross-linking. Acid-catalyzed hydrolysis of the imines **2** (THF/1 N aqueous HCl, 2:1), was followed by transformation to the resin-bound Fmoc derivatives (Fmoc-Cl, 10 equiv.; DIEA, 20 equiv.; NMP) using an in situ neutralization protocol to minimize competing intramolecular cyclization. Final cleavage (TFA/Et₃SiH, 95:5) gave the *N*^z-Fmoc protected ω -chloro- α -amino acids **4a–d** (Scheme 1 and Table 1) with HPLC purities of the crude products from 75% to 100% and purified, isolated yields from 63 to 74%. For compound **4b** ($n=3$), HPLC purity and isolated yield was lower because of the expected partial competing cyclization to form proline (25% by HPLC).

Following characterization of key intermediates **2** as described above, on-resin displacement of the ω -halide by a nucleophile to provide side-chain functionalized products **6** was investigated (Scheme 1 and Table 1). An homologous series of ω -nitrile derivatives (length of side chain $n=2-5$) was prepared by displacement of the halide intermediates **2** (Bu₄NCN and Bu₄NI, 10 equiv. each; NMP; 24 h; 25°C). For cases in which $n=3$ to 5 the intermediate **2** was the ω -chloride, while for $n=2$ the ω -bromide was used, since in this case the ω -chloride derivative was not of sufficient reactivity. Additionally, displacement of the chloride intermediate **2c** ($n=4$) with a less reactive nucleophile, benzyl mercap-

Keywords: alkylation; combinatorial chemistry; α,ω -dihaloalkanes; nucleophilic substitution; proline homologs; Schiff base activation; side-chain diversity; spirocyclic amino acids.

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Scheme 1. Synthesis of side-chain reactive unnatural amino acids, proline and homologs, and spirocyclic amino acids, from key resin-bound intermediates **2**.

Table 1. *N*^α-Fmoc α-amino acids containing ω-chloride, nitrile or thioether functionalities, prepared from key resin-bound intermediates **2**

	n = 2		n = 3		n = 4			n = 5	
	4a	6a	4b	6b	4c	6c	6c'	4d	6d
	X = Cl	X = CN ^a	X = Cl	X = CN	X = Cl	X = CN	X = SCH ₂ Ph	X = Cl	X = CN
Crude HPLC Purity	89%	50%	75%	92%	93%	97%	63%	100%	100%
Purified Yield	67%	23%	63%	63%	71%	76%	35%	74%	71%

^aThe bromide was used as the halide precursor.

tide (20 equiv.; DIEA, 20 equiv.; NMP; 85°C; 36 h) provided the thioether derivative. Hydrolysis of **5**, acylation of the free α-amine with Fmoc-Cl, and TFA cleavage yielded products **6a–d** and **6c'** with crude HPLC purities ranging from 50 to 100%, and purified yields from 23 to 76% (Table 1).

Our next goal was to obtain cyclic imino acids (**8**) of different ring sizes.⁷ Resin-bound five- to seven-membered proline ring homologs were prepared starting

from intermediate **3**, by neutralization and intramolecular halide displacement with the α-amino group (Scheme 1). Cyclization conditions varied depending on the ring size. For *n* = 3 (five-membered ring) and *n* = 4 (six-membered ring), cyclization involved using the ω-chloro derivatives (10% DIEA in NMP, 25°C, 24 h). In addition, for *n* = 5 (seven-membered ring), cyclization required using the ω-bromo derivative and higher temperature (85°C); for *n* = 2 (four-membered ring), all experiments led to formation of the *N*-alkyl cross-link-

ing product by intermolecular halide displacement by the α -amino group. Acylation of the cyclic secondary amine with Fmoc-Cl, followed by TFA cleavage gave products **8a** and **8c–8d** with excellent crude HPLC purities (88–98%) and good purified yields (55–73%) (Table 2). Following identical procedures, N^{α} -Fmoc-4-methylproline **8b** (mixture of four stereoisomers) and N^{α} -Fmoc-1,2,3,4-tetrahydroisoquinoline-3-carboxylic acid **8e** were prepared in good purities and yields (Table 2) by using, respectively, (\pm)-1-bromo-3-chloro-2-methylpropane and α,α' -dichloro-*o*-xylene as the alkylating agents, and 10% DIEA in NMP for 24 h at 25°C for the cyclization.

1-Aminocycloalkanecarboxylic acid spiro derivatives **10** can also be synthesized starting from intermediate **2**, by an intramolecular C^{α} -alkylation (Scheme 1).⁸ These conformationally constrained α -amino acid derivatives were prepared from the ω -chloro derivatives (BTPP, 10 equiv., NMP, 85°C, 24 h). Hydrolysis of **9**, followed by acylation with Fmoc-Cl, and TFA cleavage provided N^{α} -Fmoc protected 1-aminocyclopropane-1-carboxylic acid ($n=2$, **10a**), and homologs ($n=4$, **10b**, and 2-aminoindan-2-carboxylic acid, **10c**) with crude HPLC purities ranging from 72 to 100%, and purified yields from 29 to 65% (Table 2).

In summary, alkylation of the benzophenone imine of glycine Wang resin with α,ω -dihaloalkanes gives access to key ω -halo intermediates. These intermediates were transformed on-resin to α -monosubstituted amino acids containing diverse side-chain functionalities, proline and its ring homologs, and 1-aminocycloalkanecarboxylic acid derivatives. In addition, the incorporation of the ω -haloalkanes **3** and **4** into peptides could provide potential alkylating agents for applications such as affinity labeling¹⁰ and intramolecular cyclization.¹¹

Experimental

Preparation of the benzophenone imine of Gly-Wang-resin (1**).** In a 5 mL syringe,¹² Fmoc-Gly-Wang-resin (0.2 g, 1.0 mmol/g) was washed with CH_2Cl_2 and DMF (2 \times 1 min each), and then treated with piperidine–DMF (1:4, 3 \times 5 min), followed by washings with DMF (6 \times 0.5 min). Benzophenone imine (336 μL , 10 equiv.) in NMP (2 mL) was added to the resin, followed by the addition of glacial acetic acid (100 μL , 8.7 equiv.), and the reaction was allowed to proceed for 24 h with rotation. The resultant resin was washed with NMP and THF (4 \times 0.5 min each).

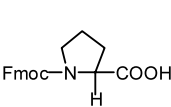
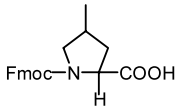
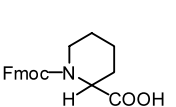
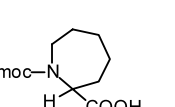
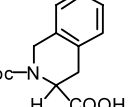
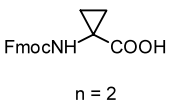
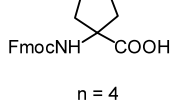
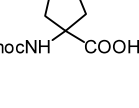
Alkylation of resin **1 with an α,ω -dihaloalkane.** Resin-bound Schiff base **1** (200 μmol) was washed with NMP (2 \times 0.5 min). The α,ω -dihaloalkane (10 equiv.) in NMP (2 mL) and BTPP (610 μL , 10 equiv.) were added, and the suspension was rotated for 24 h at 25°C. The resin was washed with NMP and CH_2Cl_2 (4 \times 0.5 min each).

Nucleophilic displacement of the halide. The resin bound imine **2** (200 μmol) was swollen with NMP (2 \times 0.5 min). Bu_4NI (925 mg, 10 equiv.) and Bu_4NCN (670 mg, 10 equiv.) were individually dissolved in NMP (1.5 mL each), combined, and added to the resin. The reaction mixture was rotated for 24 h at 25°C. The resulting resin **5** was washed with NMP and CH_2Cl_2 (6 \times 0.5 min each).

Hydrolysis of the imine. Resin-bound imine **2** or **5** (200 μmol) was washed with THF (6 \times 0.5 min). THF–1N HCl aqueous (2:1, 4 mL) was added, and the suspension was rotated for 4 h at 25°C. The resin was washed with THF and CH_2Cl_2 (4 \times 0.5 min each).

Acylation of resin-bound product with Fmoc-Cl. Resin-bound amine **3** (200 μmol) was washed with NMP

Table 2. Products from conversion of intermediates **2** to N^{α} -protected Fmoc proline, its ring homologs, and 1-aminocycloalkanecarboxylic acid derivatives

					
n = 3		n = 4		n = 5	
8a	8b	8c	8d^a	8e	
Crude HPLC Purity	98%	98%	94%	88%	95%
Purified Yield	68%	77%	73%	55%	72%
^a The bromide was used as the halide precursor.					
					
n = 2	n = 4				
10a	10b		10c		
Crude HPLC Purity	73%		72%		
Purified Yield	58%		29%		

(4×0.5 min). Fmoc-Cl (516 mg, 10 equiv.) was dissolved in NMP (2 mL), added to the resin, and acylation was started by addition of DIEA (680 μ L, 20 equiv.). Reaction mixture was rotated for 24 h at 25°C. The resin was washed with NMP, THF, and CH₂Cl₂ (4×0.5 min each).

Cleavage and purification. The resin was cleaved with TFA–Et₃SiH (95:5, 1×2 h, 1×30 min). Filtrates were collected, combined with TFA–CH₂Cl₂ washes (1:3, 2×2 min) of the resin, and evaporated. Crude products were purified over silica gel with CHCl₃–THF–HOAc (92:8:1).

6-Chloro-2-[(9H-fluoren-9-ylmethoxy)carbonyl] amino]-hexanoic acid (4c). Using 1-bromo-4-chlorobutane (230 μ L, 10 equiv.) in the alkylation step. ¹H NMR (CD₃OD) δ 1.48–1.66 (m, 2H), 1.66–2.00 (m, 4H), 3.61 (t, *J*=6.6 Hz, 2H), 4.14–4.24 (m, 1H), 4.27 (t, *J*=6.6 Hz, 1H), 4.40 (d, *J*=6.6 Hz, 2H), 7.30–7.48 (m, 4H), 7.64–7.76 (m, 2H), 7.84 (d, *J*=7.2 Hz, 2H); ¹³C NMR (CD₃OD) δ 24.3, 32.0, 33.2, 45.4, 48.6, 55.2, 67.9, 120.9, 126.2, 128.1, 128.7, 142.6, 145.2, 145.4, 158.7, 175.9.

6-Cyano-2-[(9H-fluoren-9-ylmethoxy)carbonyl]amino]-hexanoic acid (6c). Using 1-bromo-4-chlorobutane (230 μ L, 10 equiv.) in the alkylation step. ¹H NMR (CD₃OD) δ 1.50–1.65 (m, 2H), 1.65–1.85 (m, 3H), 1.85–2.00 (m, 1H), 2.49 (t, *J*=6.9 Hz, 2H), 4.15–4.25 (m, 1H), 4.27 (t, *J*=7.5 Hz, 1H), 4.41 (d, *J*=7.5 Hz, 2H), 7.30–7.52 (m, 4H), 7.62–7.78 (m, 2H), 7.84 (d, *J*=7.5 Hz, 2H); ¹³C NMR (CD₃OD) δ 17.2, 26.0, 26.1, 31.9, 48.8, 55.1, 67.9, 120.9, 121.0, 126.3, 128.2, 128.8, 142.6, 145.2, 145.3, 158.7, 175.7.

Alternative synthetic procedure for proline derivatives

Intramolecular cyclization of the resin-bound alkylated products. The resin-bound amine **3** (200 μ mol) was washed with NMP (4×0.5 min). 10% DIEA in NMP (4 mL) was added to the resin and the reaction mixture was rotated for 24 h at 25°C. The resin was washed with NMP and CH₂Cl₂ (4×0.5 min each).

1-(9H-Fluoren-9-ylmethyl) hydrogen 1,2-piperidine dicarboxylate (8c). Using 1-bromo-4-chlorobutane (230 μ L, 10 equiv.) in the alkylation step. ¹H NMR (CD₃OD, 3:2 mixture of two rotamers) δ 1.24–1.54 (m, 2H), 1.54–1.82 (m, 3H), 2.16–2.36 (m, 1H), 2.94–3.20 (m, 1H), 3.88–4.10 (m, 1H), 4.22–4.36 (m, 1H), 4.36–4.52 (m, 2H), 4.68 (d, *J*=5.1 Hz, 0.4 H), 4.83 (d, *J*=5.1 Hz, 0.6 H), 7.28–7.50 (m, 4H), 7.56–7.72 (m, 2H), 7.84 (m, 2H); ¹³C NMR (CD₃OD, mixture of two rotamers) δ 21.7, 25.7, 25.8, 27.8, 42.8, 43.0, 48.6, 55.6, 55.8, 68.8, 120.9, 126.1, 128.2, 128.8, 142.6, 145.2, 145.3, 145.4, 157.8, 158.1, 174.6.

Alternative synthetic procedure for spiro derivatives

Intramolecular cyclization by C α -alkylation of the resin-bound alkylated products. Resin **2** (200 μ mol) was washed with NMP (2×0.5 min). Spiro formation was carried out in a glass vessel by adding BTPP (612 μ L, 10 equiv.) in NMP (2 mL) to the resin. Reaction mixture was heated at 85°C for 24 h with occasional stirring. Resin was washed with NMP and CH₂Cl₂ (4×0.5 min each).

1-[(9H-Fluoren-9-ylmethoxy)carbonyl]amino]cyclopropanecarboxylic acid (10a). Using 1-bromo-2-chloroethane (166 μ L, 10 equiv.) in the alkylation step. ¹H NMR (CD₃OD+CDCl₃, 3:1 mixture of two rotamers) δ 0.94–1.04 (m, 0.5H), 1.08–1.22 (m, 1.5H), 1.34–1.44 (m, 0.5H), 1.44–1.58 (m, 1.5H), 4.26 (t, *J*=6.9 Hz, 1H), 4.37 (d, *J*=6.6 Hz, 1.5H), 4.46 (d, *J*=6.0 Hz, 0.5H), 7.26–7.48 (m, 4H), 7.69 (d, *J*=7.5 Hz, 2H), 7.81 (d, *J*=7.5 Hz, 2H); ¹³C NMR (CD₃OD+CDCl₃, mixture of two rotamers) δ 17.9, 34.8, 48.3, 67.8, 120.8, 126.1, 128.0, 128.6, 142.4, 145.1, 159.1, 176.7.

Acknowledgements

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